

Physics 742 – Graduate Quantum Mechanics 2
Solutions to Chapter 15

5. [20] A hydrogen atom is in the 1s ground state while being bathed in light of sufficient frequency to excite it to the $n = 2$ states. The light is traveling in the $+z$ direction and has circular polarization, $\boldsymbol{\varepsilon} = \frac{1}{\sqrt{2}}(\hat{x} + i\hat{y})$.

(a) [10] Calculate all relevant dipole moments \mathbf{r}_{FI} for final states $|2lm\rangle$.

The matrix elements we need are

$$\begin{aligned}\langle 2lm | \mathbf{R} \cdot \boldsymbol{\varepsilon} | 100 \rangle &= \frac{1}{\sqrt{2}} \langle 2lm | (X + iY) | 100 \rangle = \frac{1}{\sqrt{2}} \int d^3\mathbf{r} \psi_{2lm}^*(\mathbf{r}) \psi_{100}(\mathbf{r}) r \sin\theta (\cos\phi + i\sin\phi) \\ &= \frac{1}{\sqrt{2}} \int_0^\infty r^3 dr R_{2l}(r) R_{10}(r) \int d\Omega Y_l^m(\theta, \phi)^* Y_0^0(\theta, \phi) \sin\theta e^{i\phi}.\end{aligned}$$

Of course, $Y_0^0 = 1/\sqrt{4\pi}$, and comparison of the final factors with spherical harmonics makes it clear that $\sin\theta e^{i\phi} = -Y_1^1(\theta, \phi) \sqrt{8\pi/3}$, so when we put this all together, we have

$$\begin{aligned}\langle 2lm | \mathbf{R} \cdot \boldsymbol{\varepsilon} | 100 \rangle &= -\frac{1}{\sqrt{2}} \frac{1}{\sqrt{4\pi}} \sqrt{\frac{8\pi}{3}} \int_0^\infty r^3 dr R_{2l}(r) R_{10}(r) \int d\Omega Y_l^m(\theta, \phi)^* Y_1^1(\theta, \phi) \\ &= \frac{-1}{\sqrt{3}} \delta_{l1} \delta_{m1} \int_0^\infty r^3 dr R_{21}(r) R_{10}(r) = \frac{-2}{\sqrt{3} \cdot 24 a_0^5 a_0^3} \delta_{l1} \delta_{m1} \int_0^\infty r^3 dr (r e^{-r/2a_0}) e^{-r/a_0} \\ &= \frac{-1}{3\sqrt{2} a_0^4} \delta_{l1} \delta_{m1} 4! \left(\frac{2}{3} a_0\right)^5 = -\frac{128\sqrt{2}}{243} a_0 \delta_{l1} \delta_{m1}.\end{aligned}$$

Doing it by hand seemed easier than using Maple. We used orthogonality of the Y 's to do the angular integral.

(b) [4] Find a formula for the rate at which the atom makes this transition.

We now simply use equation (15.30) from the notes, to obtain

$$\Gamma(I \rightarrow F) = 4\pi^2 \alpha \hbar^{-1} \mathcal{I}(\omega_{FI}) \left(\frac{128\sqrt{2}}{243} a_0 \right)^2 = \frac{2^{17} \pi^2 \alpha a_0^2}{3^{10} \hbar} \mathcal{I}(\omega_{FI}).$$

(c) [6] What is the wavelength required for this transition? Assume at this wavelength the power is $\mathcal{I}(\lambda) = 100 \text{ W/m}^2/\text{nm}$. Find the rate at which the atom converts. (Note the footnote on p. 280)

The energy involved is $E = \frac{3}{4}(13.60 \text{ eV}) = 10.20 \text{ eV}$. We can then calculate the relevant wavelength from

$$\lambda = \frac{c}{f} = \frac{hc}{E} = \frac{1240 \text{ nm} \cdot \text{eV}}{10.20 \text{ eV}} = 121.5 \text{ nm}.$$

We then calculate the intensity per unit angular frequency as

$$\mathcal{I}(\omega) = \frac{\lambda^2}{2\pi c} \mathcal{I}(\lambda) = \frac{(121.5 \text{ nm})^2 (100 \text{ W/m}^2/\text{nm})}{2\pi (2.998 \times 10^8 \text{ m/s})(10^9 \text{ nm/m})} = 7.84 \times 10^{-13} \text{ W} \cdot \text{s/m}^2.$$

Substituting this into our expression, we find

$$\Gamma(|100\rangle \rightarrow |211\rangle) = \frac{2^{17} \pi^2 (5.29 \times 10^{-11} \text{ m})^2 (7.84 \times 10^{-13} \text{ J/m}^2)}{3^{10} (137)(1.054 \times 10^{-34} \text{ J} \cdot \text{s})} = 3.33 \text{ s}^{-1}.$$

So even with this rather weak source, an atom will undergo such a transition several times per second. The reverse rate *would* be the same, except it can be shown that spontaneous emission, a process we have not yet accounted for, is a much faster process.