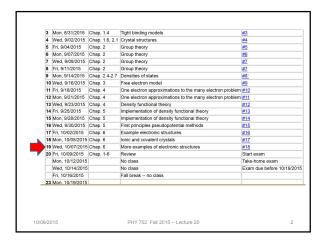
PHY 752 Solid State Physics 11-11:50 AM MWF Olin 103

Plan for Lecture 20:

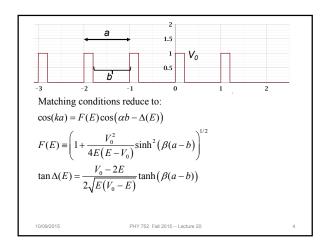
Review: Chapters 1-6 in GGGPP

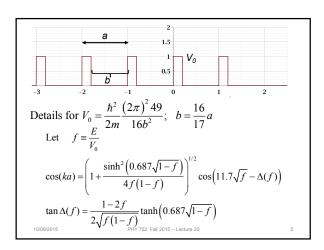
- 1. Bloch theorem
- 2. Kronig-Penny model
 3. Group theory

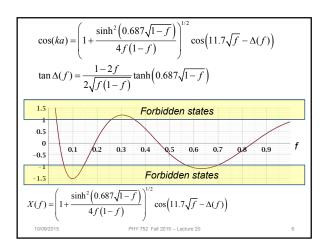
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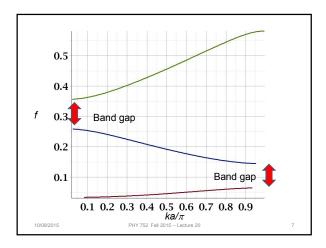


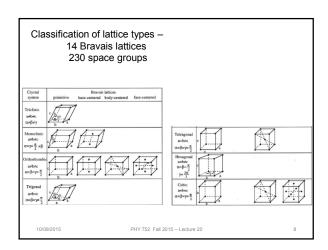
Bloch theorem for solution $\Psi_k(\mathbf{r})$ to the Schroedinger equation for an electron in a periodic solid in terms of translation vector T :				
$\begin{array}{l} \Psi_{_k}(r+T) = e^{A\cdot T} \Psi_{_k}(r) \\ \text{Consider an electron moving in a one-dimensional model} \\ \text{potential (Kronig and Penney, $Proc. Roy. Soc. (London)} \\ \textbf{130, 499 (1931)} \end{array}$				
	a	1.5		
		0.5	V _o	
-3	-2 -	-1 0	1	2
Schroedinger equation for electron:				
$\left(-\frac{\hbar^2\nabla^2}{2m} + V(x)\right)\Psi(x) = E\Psi(x)$ 10/09/2015 PHY 752 $\int_{\text{ell}}^{\infty} 2015 - \text{Lecture 20}$				











Short digression on abstract group theory What is group theory ?

A group is a collection of "elements" $-A,B,C,\ldots$ and a "multiplication" process. The abstract multiplication (\cdot) pairs two group elements, and associates the "result" with a third element. (For example $(A\cdot B=C)$.) The elements and the multiplication process must have the following properties.

- 1. The collection of elements is closed under multiplication. That is, if elements A and B are in the group and $A\cdot B=C$, element C must be in the group.
- 2. One of the members of the group is a "unit element" (E). That is, for any element A of the group, $A\cdot E=E\cdot A=A.$
- 3. For each element A of the group, there is another element A^{-1} which is its "inverse". That is $A\cdot A^{-1}=A^{-1}\cdot A=E.$
- 4. The multiplication process is "associative". That is for sequential mulplication of group elements $A,\,B,\,$ and $C,\,(A\cdot B)\cdot C=A\cdot (B\cdot C).$

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Group theory – some comments

• The elements of the group may be abstract; in general, we will use them to describe symmetry properties of our system

Representations of a group

A representation of a group is a set of matrices (one for each group element) -- $\Gamma(A)$, $\Gamma(B)$... that satisfies the multiplication table of the group. The dimension of the matrices is called the dimension of the representation.

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The great orthogonality theorem

Notation: $h \equiv \text{order of the group}$

 $R \equiv$ element of the group

 $\Gamma^{i}(R)_{\alpha\beta} \equiv i$ th representation of R

 $_{\alpha\beta}$ denote matrix indices

 $l_i \equiv$ dimension of the representation

$$\sum_{R} \left(\Gamma^{i}(R)_{\mu\nu} \right)^{*} \Gamma^{j}(R)_{\alpha\beta} = \frac{h}{l_{i}} \delta_{ij} \delta_{\mu\alpha} \delta_{\nu\beta}$$

Analysis shows: $\sum l_i^2 = h$

$$\sum l_i^2 = h$$

Note: only irreducible representations are used.

Simplified analysis in terms of the "characters" of the representations

$$\chi^{j}(R) \equiv \sum_{i=1}^{l_{j}} \Gamma^{j}(R)_{\mu\mu}$$

Character orthogonality theorem

$$\sum_{P} \left(\chi^{i}(R) \right)^{*} \chi^{j}(R) = h \delta_{ij}$$

Note that all members of a class have the same character for any given representation i.

Reciprocal lattice



real lattice

reciprocal lattice

Unit vectors of the reciprocal lattice

$$g_1 = \frac{2\pi}{\Omega} t_2 \times t_3, \quad g_2 = \frac{2\pi}{\Omega} t_3 \times t_1, \quad g_3 = \frac{2\pi}{\Omega} t_1 \times t_2, \quad \text{and} \quad \Omega = t_1 \cdot (t_2 \times t_3),$$

General reciprocal lattice vector

$$\mathbf{g}_m = m_1 \mathbf{g}_1 + m_2 \mathbf{g}_2 + m_3 \mathbf{g}_3$$

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Reciprocal lattice



real lattice

reciprocal lattice

Unit vectors of the reciprocal lattice

$$\mathbf{g}_1 = \frac{2\pi}{\Omega}\mathbf{t}_2 \times \mathbf{t}_3, \quad \mathbf{g}_2 = \frac{2\pi}{\Omega}\mathbf{t}_3 \times \mathbf{t}_1, \quad \mathbf{g}_3 = \frac{2\pi}{\Omega}\mathbf{t}_1 \times \mathbf{t}_2, \quad \text{and} \quad \Omega = \mathbf{t}_1 \cdot (\mathbf{t}_2 \times \mathbf{t}_3),$$

General reciprocal lattice vector

$$\mathbf{g}_m = m_1 \mathbf{g}_1 + m_2 \mathbf{g}_2 + m_3 \mathbf{g}_3$$

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Quantum Theory of materials

Exact Schrödinger equation:

Electronic coordinates

$$\begin{split} \mathcal{H}(\{\mathbf{r}_i\}, \{\mathbf{R}^a\}) \Psi_{av}(\{\mathbf{r}_i\}, \{\mathbf{R}^a\}) = E_{av} \Psi_{av}(\{\mathbf{r}_i\}, \{\mathbf{R}^A \text{tomic} \\ \text{coordinates} \end{split}$$
 where

$$\boldsymbol{\mathcal{H}}(\{\mathbf{r}_i\}, \{\mathbf{R}^a\}) = \boldsymbol{\mathcal{H}}^{\text{Nuclei}}(\{\mathbf{R}^a\}) + \boldsymbol{\mathcal{H}}^{\text{Electrons}}(\{\mathbf{r}_i\}, \{\mathbf{R}^a\})$$

Born-Oppenheimer approximation Born & Huang, Dynamical Theory of Crystal Lattices, Oxford (1954)



Approximate factorization:

$$\Psi_{\alpha\nu}(\{\mathbf{r}_i\}, \{\mathbf{R}^a\}) = X_{\alpha\nu}^{\text{Nuclei}}(\{\mathbf{R}^a\}) \Upsilon_{\alpha}^{\text{Electrons}}(\{\mathbf{r}_i\}, \{\mathbf{R}^a\})$$

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Quantum Theory of materials -- continued

Electronic Schrödinger equation:

$$\begin{split} \boldsymbol{\mathcal{H}}^{\text{Electrons}}(\{\mathbf{r}_i\}, \{\mathbf{R}^a\}) \Upsilon_{\alpha}^{\text{Electrons}}(\{\mathbf{r}_i\}, \{\mathbf{R}^a\}) &= U_{\alpha}(\{\mathbf{R}^a\}) \Upsilon_{\alpha}^{\text{Electrons}}(\{\mathbf{r}_i\}, \{\mathbf{R}^a\}) \\ \boldsymbol{\mathcal{H}}^{\text{Electrons}}(\{\mathbf{r}_i\}, \{\mathbf{R}^a\}) &= -\frac{\hbar^2}{2m} \sum_{i} \nabla_{i}^2 - \sum_{a,i} \frac{Z^a e^2}{|\mathbf{r}_i - \mathbf{R}^a|} + \sum_{i < j} \frac{e^2}{|\mathbf{r}_i - \mathbf{r}_j|} \end{split}$$

Nuclear Hamiltonian: (Often treated classically)

$$\mathbf{\mathcal{H}}^{\text{Nuclei}}\left(\{\mathbf{R}^{a}\}\right)X_{\alpha\nu}^{\text{Nuclei}}\left(\{\mathbf{R}^{a}\}\right) = W_{\alpha\nu}X_{\alpha\nu}^{\text{Nuclei}}\left(\{\mathbf{R}^{a}\}\right)$$

$$\mathcal{H}^{\text{Nuclei}}\left(\left\{\mathbf{R}^{a}\right\}\right) = \sum_{a} \frac{\mathbf{P}^{a2}}{2M^{a}} + U_{\alpha}\left(\left\{\mathbf{R}^{a}\right\}\right)$$



Effective nuclear interaction provided by electrons
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First consider electronic Hamiltonian

Electronic Schrödinger equation:

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contributions

Jellium model of metals

- · Nuclear potential represented by a uniform background of positive charge with charge density n_0 =Z/V (V representing volume per atom)
- Electrons represented as independent free electrons occupying free-electron states $E(k) = \frac{\hbar^2 k^2}{2m}$ for $0 \le k \le k_F$

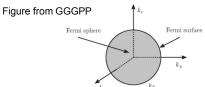
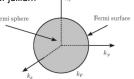


Figure 3.2 Schematic representation in the k-space of the Fermi sphere of occupied states and of the Fermi surface of the free-electron gas. At T=0 each state of wavevector ${\bf k}$, with $k< k_F$, is occupied by two electrons of either spin.

Calculation of the Fermi level for jellium



$$N = \sum_{k}^{k < k_F} 2 = 2 \frac{V}{(2\pi)^3} \frac{4}{3} \pi k_F^3 = \frac{V}{3\pi^2} k_F^3 \implies k_F^3 = 3\pi^2 r$$

.

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Density of states analysis of free electron gas

$$D(E) = 2\frac{V}{(2\pi)^3} \int \delta\left(E - \frac{\hbar^2 k^2}{2m}\right) d\mathbf{k} = \frac{V}{4\pi^3} \int_0^\infty 4\pi k^2 \, \delta\left(E - \frac{\hbar^2 k^2}{2m}\right) dk.$$

The change of variable $\hbar^2 k^2 / 2m = x$ gives

$$D(E) = \frac{V}{2\pi^2} \int_0^\infty \left(\frac{2m}{\hbar^2}\right)^{3/2} x^{1/2} \, \delta(E - x) \, dx = \frac{V}{2\pi^2} \left(\frac{2m}{\hbar^2}\right)^{3/2} E^{1/2}, \quad E > 0. \tag{3.8a}$$

$$D(E) = \frac{3}{2} \frac{N}{E_F} \left(\frac{E}{E_F}\right)^{1/2},$$

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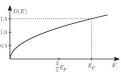


Figure 3.3 Density-of-states, in units of N/E_F , for a free-electron gas; the average electron energy (3/5) E_F is also indicated.

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Practical modeling schemes – based on density functional theory

Kohn-Sham formulation of density functional theory Results of self-consistent calculations

Variationally determined --

Ground state energy $E_{v}[n]$

Electron density $n(\mathbf{r})$

Some remaining issues

- Theory for $E_{exc}[n]$ still underdevelopment
- This formalism does not access excited states
- Strongly correlated electron systems are not well approximated

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