

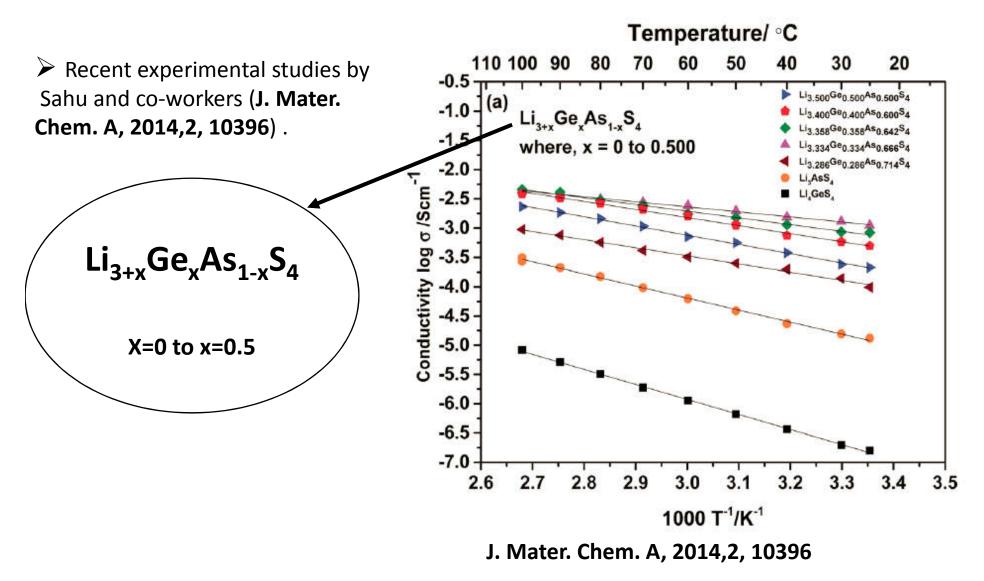
Computational study of Li ion electrolytes composed of Li₃AsS₄ alloyed with Li₄GeS₄*

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Motivation :





APS March 2016

PhysRevB.88.104103 (2013)

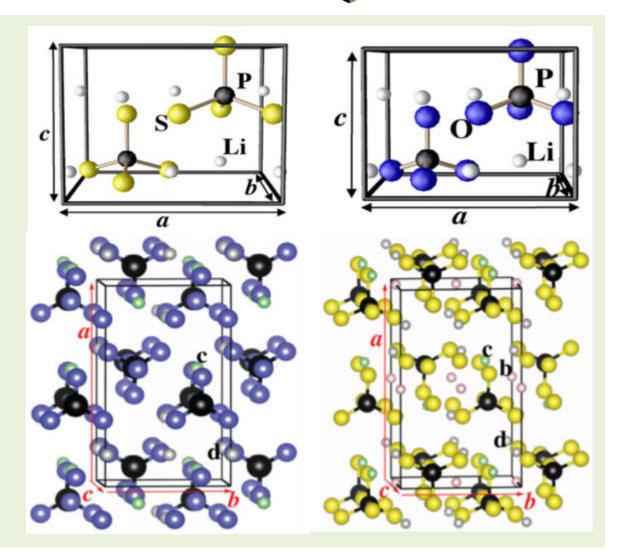
Structure of the pure Li₃AsS₄??

Pmn2₁ structure

Low temperature

Pnma structure

High temperature





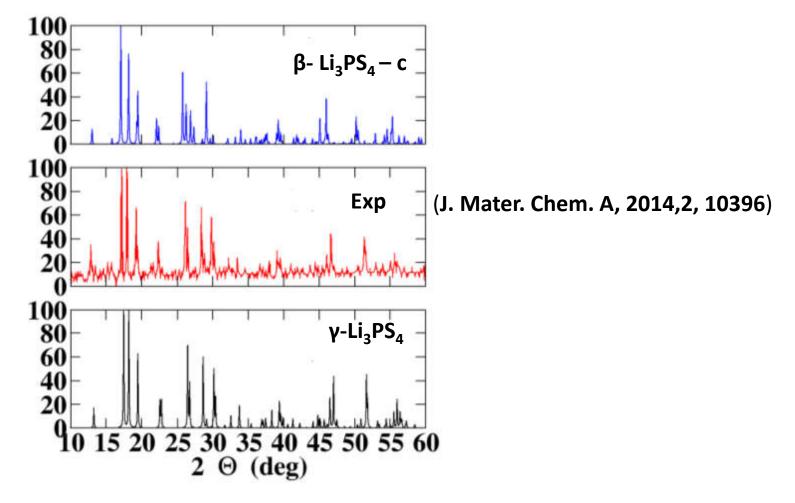


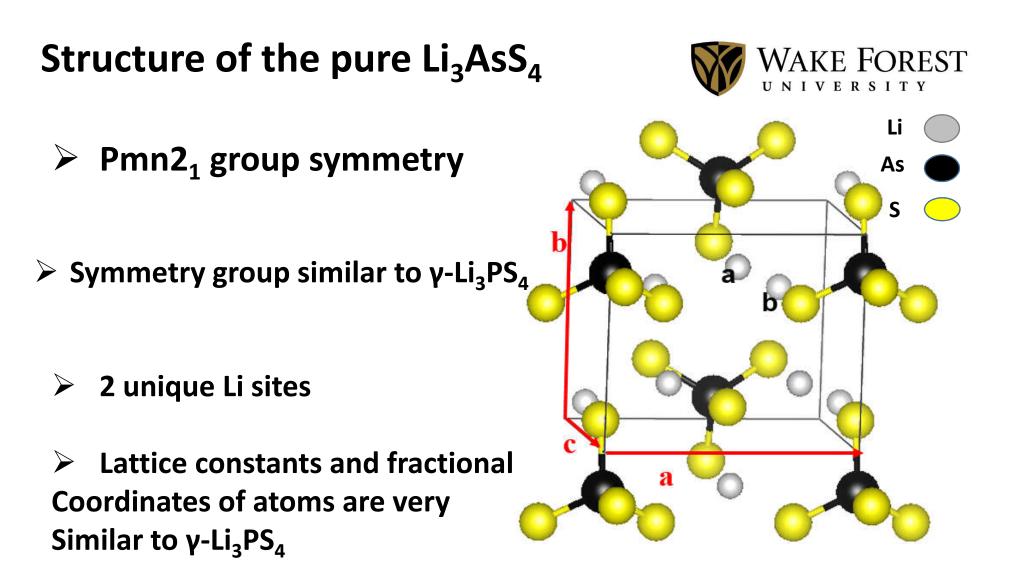
Structure of the pure Li₃AsS₄??

Structure	Li ₃ AsS ₄ Δ Η (eV)	Li ₃ PS ₄ Δ Η (eV)
γ-Li ₃ PS ₄ (Pmn2 ₁)	-7.17	-8.37
γ-Li ₃ PO ₄ (Pnma)	-6.95	-8.18
β- Li ₃ PS ₄ – b (Pnma)	-7.00	-8.28
<mark>β- Li₃PS₄ – c (Pnma)</mark>	-7.03	-8.25



Structure of the pure Li₃AsS₄





Structure of the pure Li₄GeS₄

 \succ The crystal structure of **Li_aGeS**_a

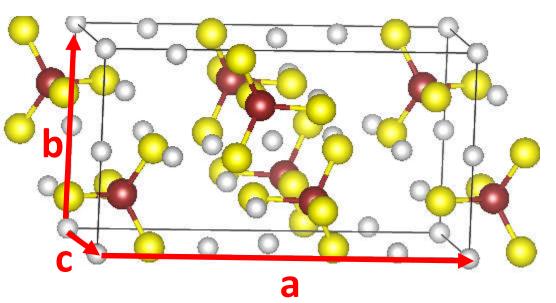


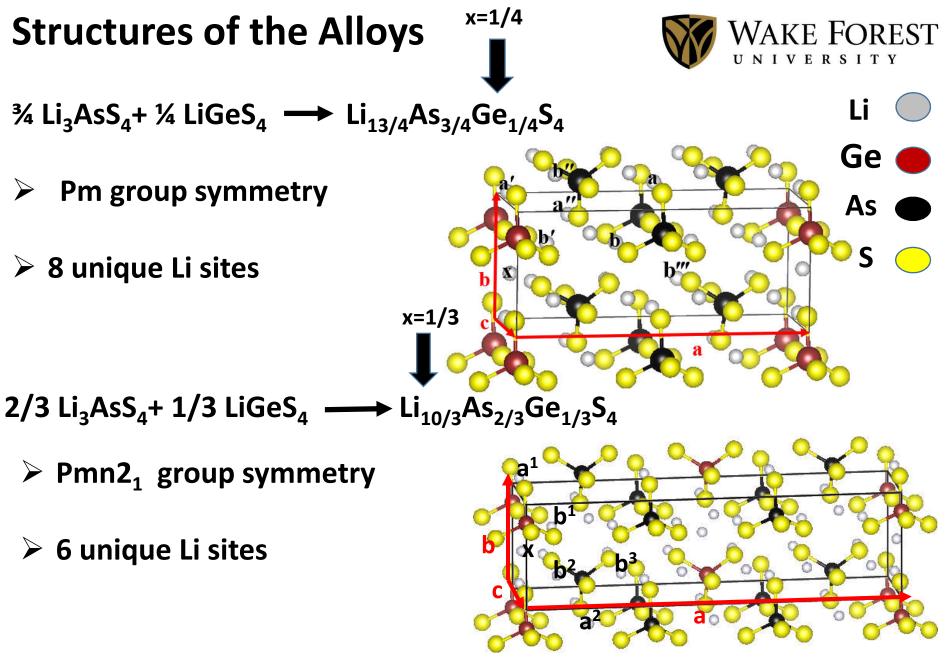
Li



Pnma space group

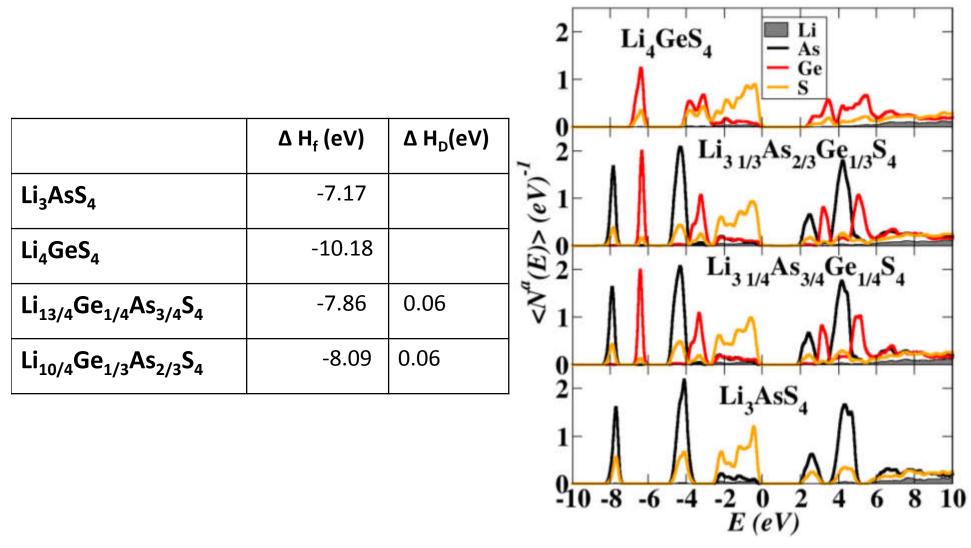
The GeS₄ tetrahedra are very similar to those AsS₄ consistent with the formation of substitutional alloys .





PDOS and HOF



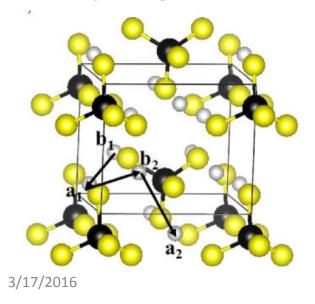


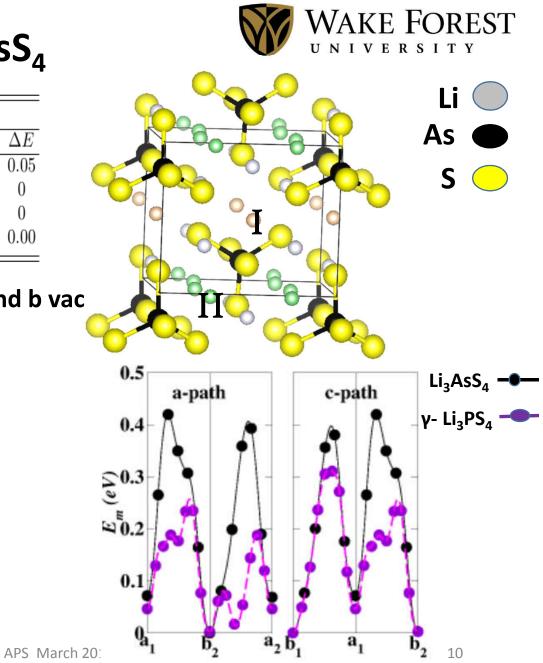
Ion migration in Li₃AsS₄

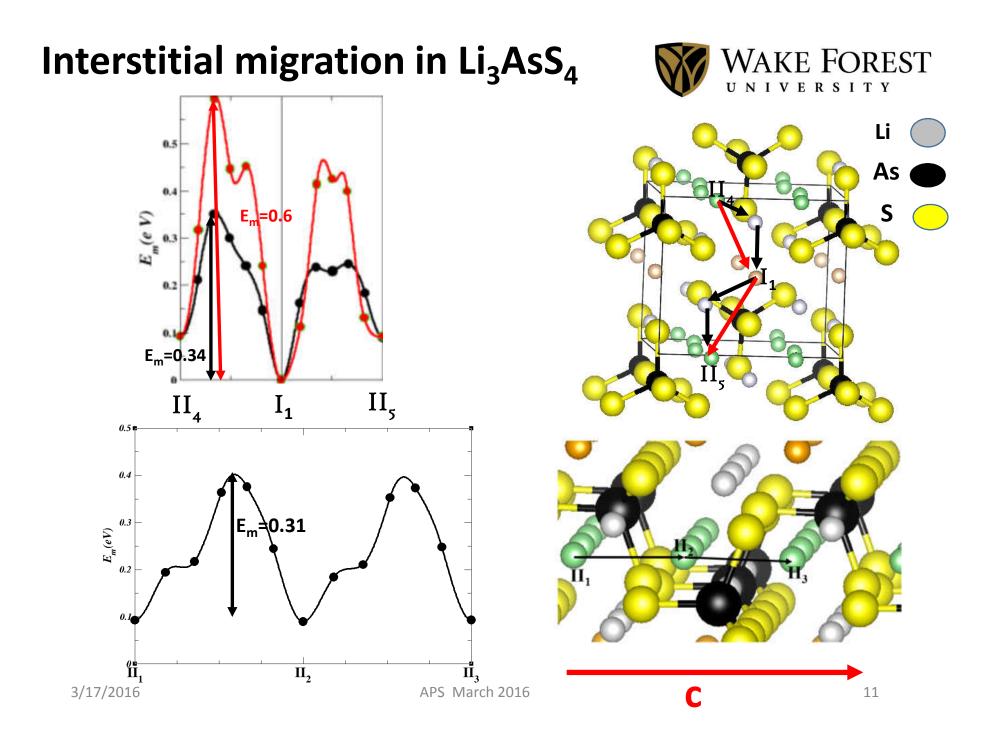
		$\rm Li_3AsS_4$		γ -Li ₃ PS ₄	
Type	Label	Position	ΔE	Position	ΔE
Vac	a	(0.00, 0.85, 0.99)	0.07	(0.00, 0.82, 1.00)	0.05
Vac	b	(0.24, 0.32, 0.00)	0	(0.24, 0.32, 0.00)	0
Inter	Ι	(0.00, 0.49, 0.65)	0	(0.00, 0.48, 0.64)	0
Inter	II	(0.23, 0.00, 0.68)	0.09	(0.24, 0.00, 0.66)	0.00

 \succ E_f = 0.85 eV by forming I inter and b vac

Vacancy migration

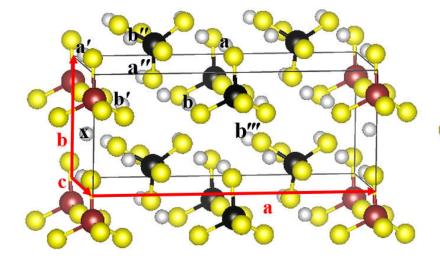


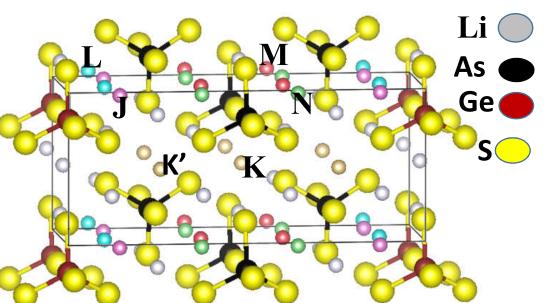




Ion migration in $Li_{13/4}Ge_{1/4}As_{3/4}S_4$







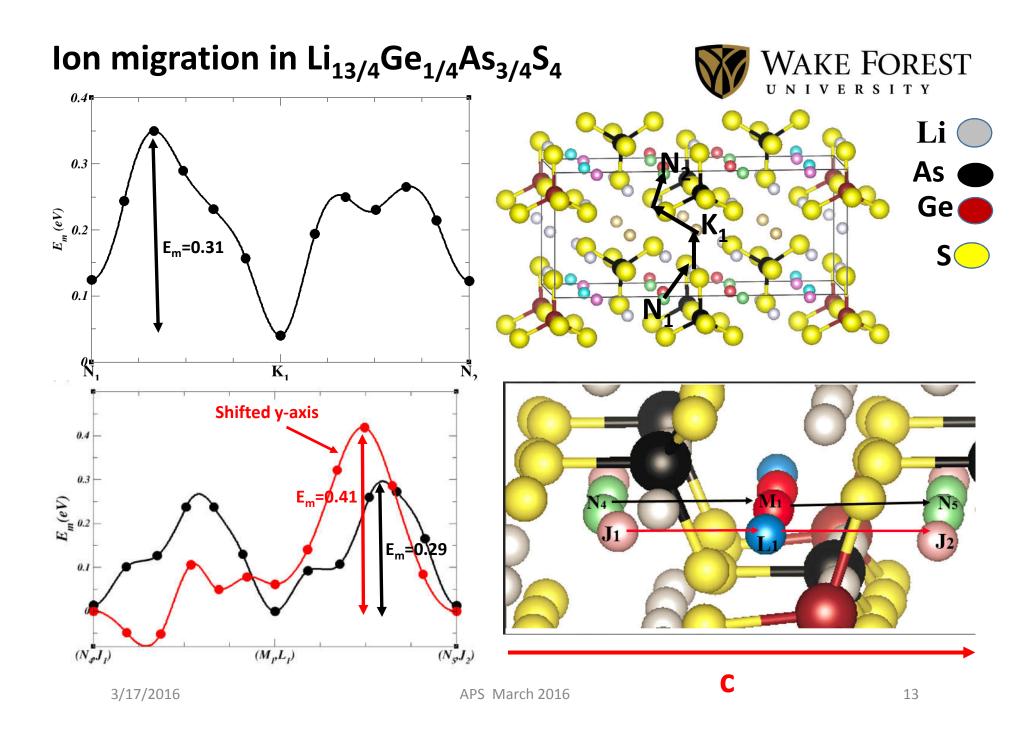
Type	Label	Ref	proximity (A^{o})	ΔE
vac	а	a	8.51	0.61
vac	*a [′]	a	3.58	0.00
vac	a″	a	7.07	0.67
vac	b	b	6.52	0.56
vac	*b [′]	b	3.75	0.01
vac	*b ^{″′}	b	3.77	0.01
vac	b‴	b	6.60	0.59
vac	x		3.01	0.01
Inter	J	II	4.52	0.0
Inter	L	II	2.62	0.07
Inter	Ν	II	7.20	0.13
Inter	М	II	5.94	0.12
Inter	K	Ι	8.81	0.06
Inter	K [′]	Ι	4.27	0.06

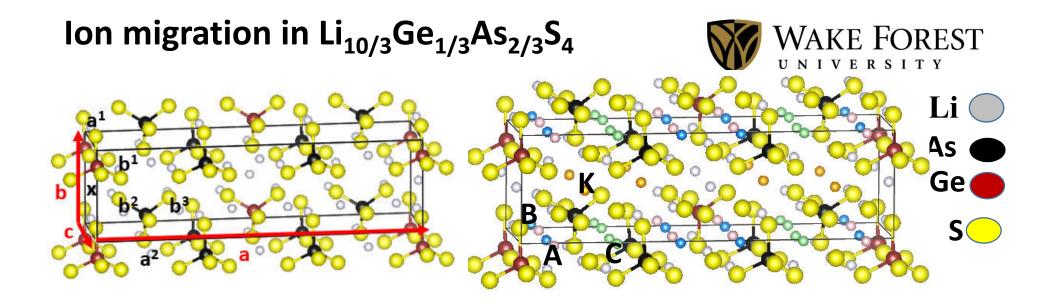
➢ E_f=0.21 eV

Involve the formation the J interstitial and x vacancy.

*One interesting result of the calculations is that vacancies close to the Ge atom (a', b', b'') are unstable; they relax to an x site vacancy.

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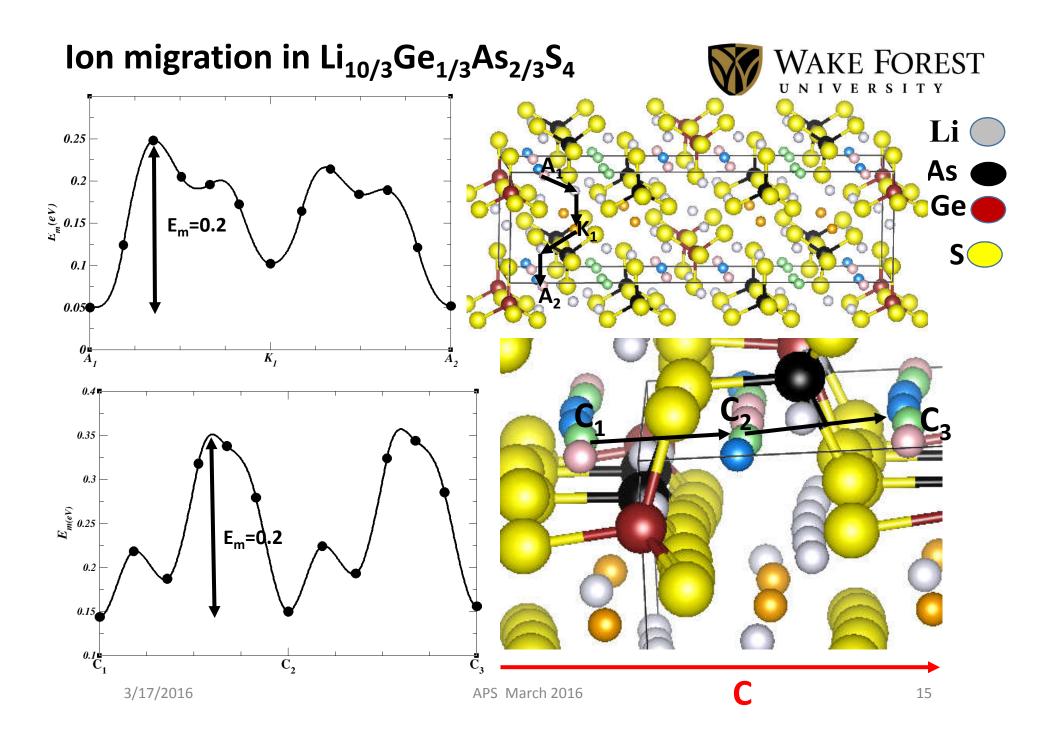


Type	Label	Ref	proximity (\mathbf{A}^{o})	ΔE
vac	$*a^1$	a	3.57	0.02
vac	a^2	a	3.85	0.62
vac	$\mathbf{*b}^{1}$	b	3.75	0.00
vac	$\mathbf{*b}^2$	b	3.78	0.01
vac	b^3	b	6.54	0.51
vac	x		3.00	0.01
Inter	А	II	2.62	0.05
Inter	В	II	2.95	0.0
Inter	С	II	5.92	0.15
Inter	Κ	Ι	4.32	0.10

E_f=0.11 eV

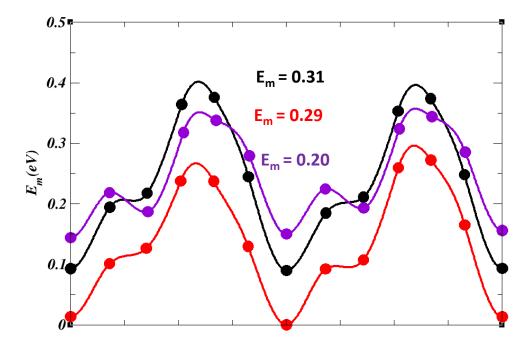
Which involves the formation ofB interstitial and x vacancy.

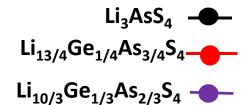
*One interesting result of the calculations is that vacancies close to the Ge atom (a¹, b¹, b²) are unstable; they relax to an x site vacancy.

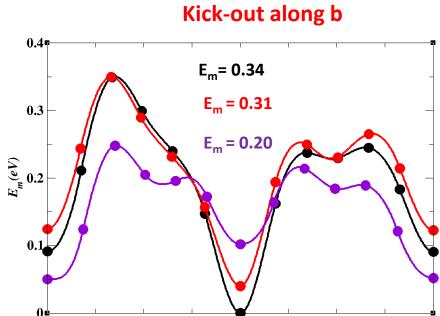


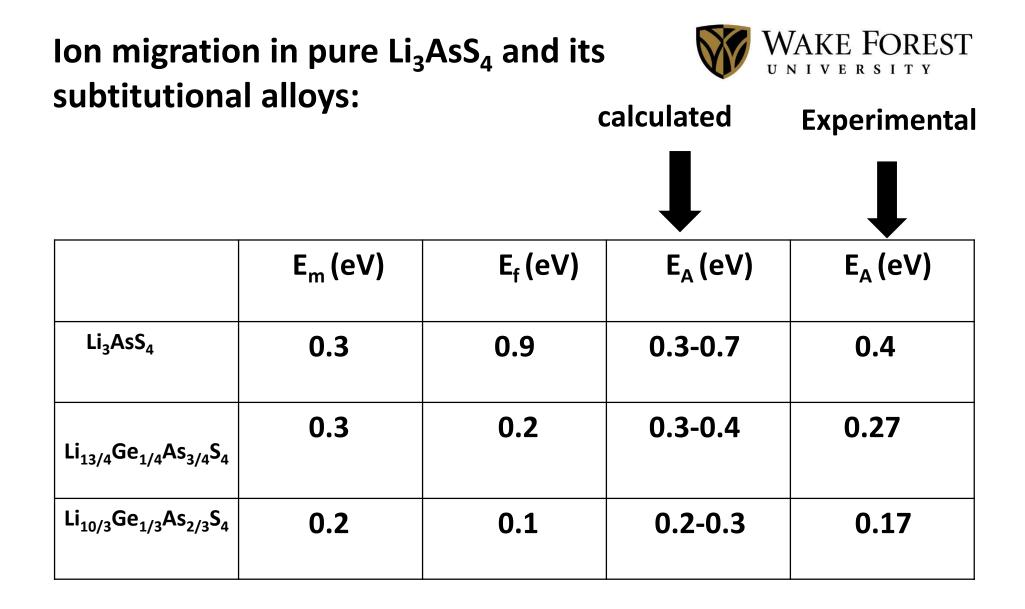
Ion migration in pure Li₃AsS₄ and its subtitutional alloys:

Interstitial migration along c









Summary and conclusions:



- In this study we report a likely structure for Li₃AsS₄ to be characterized by the Pmn2₁ space group, consistent with the X-ray measurements of Sahu and coworkers (J. Mater. Chem. A, 2014,2, 10396).
- Plausible ordered structures for substitutional alloys of Li₃AsS₄ with Li₄GeS₄ -- (1-x)Li₃AsS₄+xLi₄GeS₄ are found for x=1/4 and x=1/3.
- > The partial densities of states for the x=1/4 and x=1/3 alloys are very similar to each other and the band gaps are similar to that of Li_3AsS_4 .
- \blacktriangleright Li ion migration in Li₃AsS₄ is found to involve interstitial sites, with likely diffusion paths involving pure interstitial or kick-out mechanisms.
- Li ion migration pathways in the x=1/4 and x=1/3 alloys are found to be similar to those of Li₃AsS₄ but with lower barriers.
- The estimated activation energies for Li ion migration are consistent with the experimental trend.

3/17/2016



Computational methods

- Density functional theory with LDA
- PAW formalism using datasets generated with ATOMPAW code (Holzwarth et al. CPC 135, 329 (2001)) <u>http://pwpaw.wfu.edu</u>
- Electronic structure calculations performed using QUANTUM ESPRESSO and ABINIT codes. (Giannozzi et al. JPCM 21, 394402 (2009); <u>http://www.quantum-espresso.org,</u> Gonze et al., CPC 180, 2582 (2009))
- Plane wave expansion for wave functions with $|\mathbf{k} + \mathbf{G}|^2 \le 64 \text{ Ry}$
- Brillouin zone integration mesh of 0.003 bohr⁻³
- Ion migration energies estimated with Nudged Elastic Band (NEB) method. (Hinkleman et al. JCP 113, 9901 & 9978 (2000))
- Visualization software: Xcrysden, VESTA
- X-ray powder diffraction simulated using *Mercury*



Thank You