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### A formalism for modeling solid electrolyte/electrode interfaces using first principles methods\*

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Ref: N. D. Lepley and N. A. W. Holzwarth, **PRB 92**, 214201 (2015) N. D. Lepley, *Ph. D Thesis*, Wake Forest University, 2015.

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**FOCUS**: Solid state batteries – understanding interfaces between electrolytes and electrodes

- > Develop and analyze a quantitative measure of interface energy, adjusting for the effects of lattice strain  $\tilde{\gamma}_{ab}^{int}$  which is particularly helpful for studying (meta-)stable interfaces
- Analyze the relationship of bulk and interface stability of Li<sub>3</sub>PO<sub>4</sub>/Li and Li<sub>3</sub>PS<sub>4</sub>/Li interfaces
- Analyze the effects of charge transfer across an interface

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#### **Ideal interface**



#### **Strained interface**



Interacting interface



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Calculational methods:

- Density functional theory with LDA (Perdew, PRB 45, 13244 (1992))
- Projector Augmented Wave method (Blöchl, PRB 50, 17953 (1994))
- ATOMPAW atomic datasets (Holzwarth, CPC **135**, 329 (2001))
- Quantum Espresso code (Giannozzi, J. Phy.:CM **21**, 395502 (2009))
- Plane wave cut off:  $|\mathbf{k} + \mathbf{G}|^2 \le 64 \text{ bohr}^{-2}$
- Nudged elastic band simulations (Henkelman, JCP **113**, 9978 (2000)
- Visualization software: Kokalj, Comp. Mater. Sci. 28, 155 (2003) and Momma, Acta Cryst. 44, 1272 (2011)

It is convenient to model the interface between a solid electrolyte and solid electrode in the slab geometry using a periodic simulation cell:





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Within any given periodic simulation cell with  $n_a$  units of material a and with  $n_b$  units of material b, we can define an interface energy:



In order approximately remove the effects of lattice strain: • Design the supercell to be commenserate with lattice *a* • Now the strain will scale with the amount of material *b*  $\Rightarrow \tilde{\gamma}_{ab} \left( \tilde{\Omega}, n_a, n_b \right) = \tilde{\gamma}_{ab}^{\lim} \left( \tilde{\Omega} \right) + n_b \sigma$ 





System	${\widetilde \gamma}^{ m lim}_{ab}$ (meV/Ų)	σ (meV/Ų)
$Li_{2}O[110]/Li(\Omega_{1})$	30	6.1
Li <sub>2</sub> O[110]/Li(Ω <sub>2</sub> )	26	0.2
Li <sub>2</sub> S[110]/Li(Ω <sub>3</sub> )	19	0.2
$Li_2S[100]/Li(\Omega_4)$	19	0.0
$\gamma-{\sf Li}_3{\sf PO}_4$ [010]/Li( $\Omega_3$ )	31	0.0
γ–Li <sub>3</sub> PS <sub>4</sub> [010]/Li <sub>2</sub> S [110]	16	1.0
γ–Li <sub>3</sub> PS <sub>4</sub> [010]/Li	-216	-0.1



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#### Stable interface; composite electrolyte system

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 $Li_3PS_4 + 8Li \longrightarrow Li_3P + 4Li_2S + 12.30 eV$ 



## Partial density of states analysis of unstable $Li_3PS_4/Li$ interface:





Bulk reactions from estimated heats of formation

## $Li_3PS_4 + 8Li \longrightarrow Li_3P + 4Li_2S + 12.30 eV$ Decomposition at interface

# $Li_3PO_4 + 8Li \longrightarrow Li_3P + 4Li_2O + 6.64 eV$ (Meta-)stable interface

### Evidence of kinetic barrier at Li<sub>3</sub>PO<sub>4</sub>/Li interface





### Modeling of charge transfer in $Li_3PO_4/Li$ system





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### **Summary and conclusions**

- > A practical scheme was developed to compute an intensive measure of the interface interaction  $\tilde{\gamma}_{ab}^{int}$ , explicitly accounting for the effects of lattice stain.
- Discussed bulk reactivity as related to the interface stability of the interfaces of
  - Li<sub>3</sub>PO<sub>4</sub>/Li (having a significant kinetic barrier to decomposition)
  - $\succ$  Li<sub>3</sub>PS<sub>4</sub>/Li (having localized decomposition).
- Discussed effects of charge transfer across Li<sub>3</sub>PO<sub>4</sub>/Li interfaces – small supercells result in large internal electrostatic fields