

**PHY 742 Quantum Mechanics II**  
**1-1:50 AM MWF Olin 103**

**Plan for Lecture 7**

**Approximation methods for analyzing quantum mechanical systems**

**1. Start reading Chapter 12-13**  
**a. Variational methods**

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**Topics for Quantum Mechanics II**

**Single particle analysis**  
 Single particle interacting with electromagnetic fields – EC Chap. 9  
 Scattering of a particle from a spherical potential – EC Chap. 14  
**More time independent perturbation methods – EC Chap. 12, 13**  
 Single electron states of a multi-well potential → molecules and solids – EC Chap. 2,6  
 Time dependent perturbation methods – EC Chap. 15  
 Path integral formalism (Feynman) – EC Chap. 11.C  
 Relativistic effects and the Dirac Equation – EC Chap. 16

**Multiple particle analysis**  
 Quantization of the electromagnetic fields – EC Chap. 17  
 Photons and atoms – EC Chap. 18  
 Multi particle systems; Bose and Fermi particles – EC Chap. 10  
 Multi electron atoms and materials  
 Hartree-Fock approximation  
 Density functional approximation

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**PHY 742 Quantum Mechanics II>**

**MWF 1-1:50 PM | OPL 103 | <http://www.wfu.edu/~natalie/s20phy742/>**

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**Course schedule for Spring 2020**  
 (Preliminary schedule – subject to frequent adjustment.)

#	Lecture date	Reading	Topic	HW	Due date
1	Mon: 01/13/2020	Chap. 9	Quantum mechanics of electromagnetic forces	#1	01/22/2020
2	Wed: 01/15/2020	Chap. 9	Quantum mechanics of particle in electrostatic field	#2	01/24/2020
3	Fri: 01/17/2020	Chap. 9	Quantum mechanics of particle in magnetostatic field	#3	01/27/2020
	Mon: 01/20/2020	No class	Martin Luther King Holiday		
4	Wed: 01/22/2020	Chap. 14	Scattering theory	#4	01/29/2020
5	Fri: 01/24/2020	Chap. 14	Scattering theory	#5	01/31/2020
6	Mon: 01/27/2020	Chap. 14	Scattering theory	#6	02/03/2020
7	Wed: 01/29/2020	Chap. 12	Variational methods	#7	02/05/2020
8	Fri: 01/31/2020				
9	Mon: 02/03/2020				
10	Wed: 02/05/2020				
11	Fri: 02/07/2020				

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Variational methods for estimating the lowest energy eigenstate of a quantum mechanical system

Time independent Schrödinger equation:

$$H(\mathbf{r})\Psi(\mathbf{r}) = E\Psi(\mathbf{r})$$



Hermitian operator representing the Hamiltonian of the system

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Consider a Hamiltonian  $H$  having lowest eigenvalue  $E_0$  :

It can be shown that for any function  $\psi$

$$\frac{\langle \psi | H | \psi \rangle}{\langle \psi | \psi \rangle} \geq E_0$$

Proof: The Hamiltonian has a complete set of eigenvalues and eigenvectors:  $H|\phi_i\rangle = E_i|\phi_i\rangle$

Expanding  $|\psi\rangle$  in eigenvector basis:  $|\psi\rangle = \sum_i C_i |\phi_i\rangle$

$$\frac{\langle \psi | H | \psi \rangle}{\langle \psi | \psi \rangle} = \frac{\sum_i |C_i|^2 E_i}{\sum_i |C_i|^2} \geq E_0$$

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Significance of this inequality --

$$\frac{\langle \psi | H | \psi \rangle}{\langle \psi | \psi \rangle} \geq E_0$$

The inequality motivates a class of estimation methods known as variational methods to converge to the ground state energy  $E_0$  and the corresponding ground state probability amplitude.

Define  $E_{\text{trial}}(\Psi_{\text{trial}}) \equiv \frac{\langle \Psi_{\text{trial}} | H | \Psi_{\text{trial}} \rangle}{\langle \Psi_{\text{trial}} | \Psi_{\text{trial}} \rangle}$

Minimize  $E_{\text{trial}}(\Psi_{\text{trial}})$  with respect to  $\Psi_{\text{trial}}$

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Example --

Consider the case of a He ( $Z=2$ ) atom:

$$H(\mathbf{r}_1, \mathbf{r}_2) = -\frac{\hbar^2}{2m}(\nabla_1^2 + \nabla_2^2) - Ze^2\left(\frac{1}{r_1} + \frac{1}{r_2}\right) + \frac{e^2}{|\mathbf{r}_1 - \mathbf{r}_2|}$$

Unlike the case of a H atom, this Hamiltonian cannot be solved analytically.

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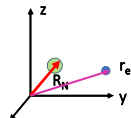
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Review -- Quantum mechanics of the hydrogen atom



$$\left(-\frac{\hbar^2}{2M}\nabla_{\mathbf{R}_N}^2 - \frac{\hbar^2}{2m}\nabla_{\mathbf{r}_e}^2 - \frac{Ze^2}{|\mathbf{r}_e - \mathbf{R}_N|}\right)\Psi(\mathbf{R}_N, \mathbf{r}_e) = E_T\Psi(\mathbf{R}_N, \mathbf{r}_e)$$

In center of mass system:

$$\mathbf{R} = \frac{M\mathbf{R}_N + m\mathbf{r}_e}{M+m} \quad \mathbf{r} = \mathbf{r}_e - \mathbf{R}_N \quad \mu = \frac{mM}{m+M}$$

$$\left(-\frac{\hbar^2}{2(M+m)}\nabla_{\mathbf{R}}^2 - \frac{\hbar^2}{2\mu}\nabla_{\mathbf{r}}^2 - \frac{Ze^2}{r}\right)\Psi(\mathbf{R}, \mathbf{r}) = E_T\Psi(\mathbf{R}, \mathbf{r})$$

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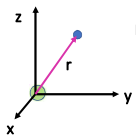
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Quantum mechanics of the hydrogen atom



In center of mass coordinates

$$\text{Reduced mass: } \mu = \frac{mM}{m+M}$$

$$\text{typically } \frac{m}{M} < \frac{1}{2000}$$

$$\left(-\frac{\hbar^2}{2\mu}\nabla_{\mathbf{r}}^2 - \frac{Ze^2}{r}\right)\psi(\mathbf{r}) = E_T\psi(\mathbf{r})$$

$$\psi(\mathbf{r}) = R_{El}(r)Y_{lm}(\hat{\mathbf{r}})$$

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Differential equation for radial wavefunction:

$$\left( -\frac{\hbar^2}{2\mu} \left[ \frac{1}{r^2} \frac{d}{dr} r^2 \frac{d}{dr} - \frac{l(l+1)}{r^2} \right] - \frac{Ze^2}{r} \right) R_{El}(r) = ER_{El}(r)$$

Convenient coordinate change

$$\rho \equiv \frac{\sqrt{8\mu|E|}}{\hbar} r \quad \text{let } \lambda = \frac{Ze^2}{\hbar} \sqrt{\frac{\mu}{2|E|}}$$

$$\left( \frac{1}{\rho^2} \frac{d}{d\rho} \rho^2 \frac{d}{d\rho} - \frac{l(l+1)}{\rho^2} + \frac{\lambda}{\rho} - \frac{1}{4} \right) R_{El}(\rho) = 0$$

For bound states,  $E = -\varepsilon$  where  $\varepsilon > 0$ .

Try solution of the form:  $R(\rho) = e^{-\rho/2} F(\rho)$

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$$\left( \frac{1}{\rho^2} \frac{d}{d\rho} \rho^2 \frac{d}{d\rho} - \frac{l(l+1)}{\rho^2} + \frac{\lambda}{\rho} - \frac{1}{4} \right) R_{El}(\rho) = 0$$

For bound states,  $E = -\varepsilon$  where  $\varepsilon > 0$ .

Try solution of the form:  $R(\rho) = e^{-\rho/2} F(\rho)$

$$\frac{d^2 F}{d\rho^2} + \left( \frac{2}{\rho} - 1 \right) \frac{dF}{d\rho} + \left( \frac{\lambda - 1}{\rho} - \frac{l(l+1)}{\rho^2} \right) F(\rho) = 0$$

Let  $F(\rho) = \rho^l L(\rho)$

$$\rho \frac{d^2 L}{d\rho^2} + (2(l+1) - \rho) \frac{dL}{d\rho} + (\lambda - l - 1)L = 0$$

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$$\rho \frac{d^2 L}{d\rho^2} + (2(l+1) - \rho) \frac{dL}{d\rho} + (\lambda - l - 1)L = 0$$

Suppose  $\lambda - l - 1 = n'$  where  $n' \geq 0$

$$\rho \frac{d^2 L}{d\rho^2} + (2(l+1) - \rho) \frac{dL}{d\rho} + n'L = 0$$

Associated Laguerre polynomial  $L_q^p(x)$ :

$$x \frac{d^2 L_q^p}{dx^2} + (p+1-x) \frac{dL_q^p}{dx} + (q-p)L_q^p = 0$$

For non-negative integers  $q$  and  $p$ .

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$$R(\rho) = \rho^l e^{-\rho/2} L_{n'-l-1}^{2l+1}(\rho)$$

Let  $n \equiv n' + l + 1 = \lambda$

$$R(\rho) = \rho^l e^{-\rho/2} L_{n-l}^{2l+1}(\rho)$$

Corresponding energy eigenvalue:

$$\lambda = \frac{Ze^2}{\hbar} \sqrt{\frac{\mu}{2|E|}} = n$$

$$\Rightarrow E = -\frac{Z^2 e^4 \mu}{2\hbar^2 n^2} \equiv -\frac{Z^2 e^2}{2a_0} \frac{1}{n^2}$$

Defining  $a_0 \equiv \frac{\hbar^2}{\mu e^2}$

Bohr radius:

$$a_0 \equiv \frac{\hbar^2}{me^2} = 0.529\,177\,210\,67 \times 10^{-10} \text{ m}$$

$$\rho = \frac{2Z}{na_0} r$$

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Example of normalized radial functions  $R_{nl}(r)$ :

$$R_{10}(r) = \left(\frac{Z}{a_0}\right)^{3/2} 2e^{-Zr/a_0}$$

$$R_{20}(r) = \left(\frac{Z}{2a_0}\right)^{3/2} \left(2 - \frac{Zr}{a_0}\right) e^{-Zr/(2a_0)}$$

$$R_{21}(r) = \left(\frac{Z}{2a_0}\right)^{3/2} \frac{Zr}{\sqrt{3}a_0} e^{-Zr/(2a_0)}$$

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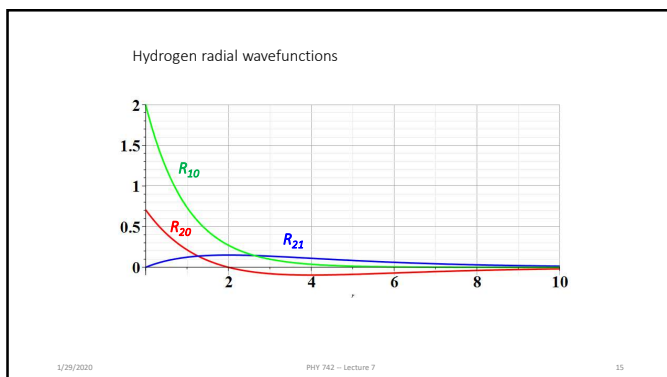
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PERIODIC TABLE  
Atomic Properties of the Elements

NIST  
National Institute of Standards and Technology  
Physical Reference Data Laboratory  
Standard Reference Data Collection

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Variational method for estimating ground state energy of a H atom:

Define  $f(\psi) \equiv \frac{\langle \psi | H | \psi \rangle}{\langle \psi | \psi \rangle}$       $\min_{\psi} f(\psi) \geq E_0$

Example:  $H(\mathbf{r}) = -\frac{\hbar^2}{2\mu} \nabla_r^2 - \frac{Ze^2}{r}$

Try:  $|\psi\rangle = e^{-\alpha r}$

$f(\psi) = \frac{\hbar^2}{2\mu} \alpha^2 - Ze^2 \alpha$

$\frac{df}{d\alpha} = 0 \Rightarrow \alpha = \frac{Ze^2 \mu}{\hbar^2} = \frac{Z}{a_0} \Rightarrow \min_{\psi} f(\psi) = -\frac{Z^2 e^2}{2a_0}$

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Variational methods for estimating ground state energy  
-- continued

Example:  $H(\mathbf{r}) = -\frac{\hbar^2}{2\mu} \nabla_r^2 - \frac{Ze^2}{r}$

Try:  $|\psi\rangle = e^{-\alpha r^2}$

$f(\psi) = \frac{\hbar^2}{2\mu} 6\alpha - \frac{4Ze^2 \sqrt{\alpha}}{\sqrt{\pi}}$

$\frac{df}{d\alpha} = 0 \Rightarrow \alpha = \frac{4Z^2}{9a_0^2 \pi} \Rightarrow \min_{\psi} f(\psi) = -\frac{Z^2 e^2}{2a_0} \frac{8}{3\pi}$

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Back to estimate of wavefunction for He atom

Consider the case of He ( $Z=2$ ):

$$H(\mathbf{r}_1, \mathbf{r}_2) = -\frac{\hbar^2}{2m}(\nabla_1^2 + \nabla_2^2) - 2e^2\left(\frac{1}{r_1} + \frac{1}{r_2}\right) + \frac{e^2}{|\mathbf{r}_1 - \mathbf{r}_2|}$$

Trial function for this case:  $\psi = \frac{Z^3}{\pi a_0^3} e^{-Z(r_1 + r_2)/a_0}$

Here  $Z$  is a variational parameter

$$f(Z) = \frac{e^2}{a_0} \left( Z^2 - \frac{27}{8} Z \right)$$

$$\frac{df}{dZ} = 0 \Rightarrow Z_{opt} = \frac{27}{16} \quad \min_{\psi} f(\psi) = -\frac{e^2}{a_0} \left( \frac{27}{16} \right)^2$$

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Variational methods for estimating ground state energy:

Define  $f(\psi) \equiv \frac{\langle \psi | H | \psi \rangle}{\langle \psi | \psi \rangle}$        $\min_{\psi} f(\psi) \geq E_0$

$$f(Z) = \frac{e^2}{a_0} \left( Z^2 - \frac{27}{8} Z \right) \quad \min_{\psi} f(\psi) = -\frac{e^2}{a_0} \left( \frac{27}{16} \right)^2$$

$$\frac{df}{dZ} = 0 \Rightarrow Z_{opt} = \frac{27}{16} \quad = -\frac{e^2}{2a_0} 5.695$$

Experimental

$$\text{value} \approx -\frac{e^2}{2a_0} 5.807$$

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